

## Product Information & Manual

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### Cholesterol esterase (CE)

Cat no. LDG0028RG

#### Product Overview

#### Specification

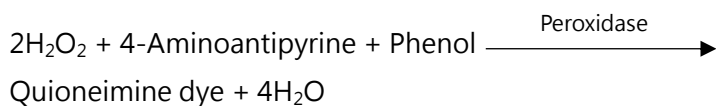
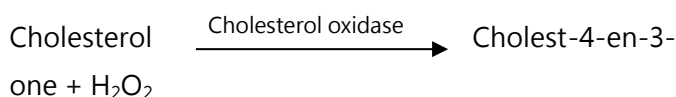
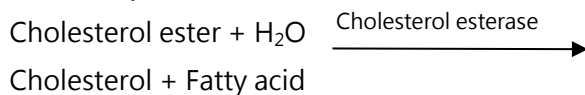
Appearance	White amorphous powder, lyophilized
Activity	70 U/mg or more

#### Properties

Stability	Stable at -20°C for at least one year
Molecular weight	52 kDa
Isoelectric point	5.5

#### Assay

##### 1. Principle



The appearance of quinoneimine dye formed when coupled with 4-aminoantipyrine, and phenol is measured at 505 nm by spectrophotometry.

##### 2. Definition

One unit causes the formation of one micromole of hydrogen peroxide (half a micromole of quinoneimine dye) per minute under the detailed conditions below.

##### 3. Reagents

<b>A. Enzyme diluent</b>	Prepare the buffer containing 20 mM K <sub>2</sub> HPO <sub>4</sub> and adjust the pH value to 7.5 with KOH. Then, add 0.019 g of MgCl <sub>2</sub> , 0.0186 g of EDTA-2Na, 0.2 g of BSA, and 0.1 g of sodium azide to the buffer, and adjust the pH value to exactly 7.0 with KOH. Fill up the solution to 100 mL with distilled water.
<b>B. Reagent 1</b>	Add 1.4 g of KH <sub>2</sub> PO <sub>4</sub> to 80 mL of distilled water. Then, add 0.0961 g of 4-Aminoantipyrine (4-AA), 1500 U of Peroxidase (POD), 3000 U of Cholesterol oxidase, and 0.1 g of sodium azide. Adjust the pH value to exactly 7.0 with KOH. Fill up the solution to 100 mL with distilled water.

<b>C. Reagent 2</b>	Add 1.4 g of $\text{KH}_2\text{PO}_4$ to 80 mL of distilled water. Then, add 0.6 g of phenol, and 0.2 g of sodium azide. Use KOH to adjust the pH value to exactly 7.0. Fill up the solution to 100 mL with distilled water.
<b>D. Reagent 3</b>	Add 1 g of Triton X-100 to 80 mL of distilled water, and heat to 70°C. Then, add 39 mg of cholesterol ester (dissolved in 2 mL isopropanol), and stir to dissolve. Stop heating until the solution becomes clear in 3-5 minutes. Add 0.6 g of sodium tauroglycocholate as the solution cold down to 60°C. Fill up the solution to 100 mL with distilled water.

#### 4. Procedure

- (1) Prepare the following working solution (for 10 tests) in a brownish bottle.

##### Working solution

<b>Reagent 1 (B)</b>	3 mL
<b>Reagent 2 (C)</b>	4.5 mL
<b>Reagent 3 (D)</b>	4.5 mL
<b>Total</b>	12 mL

- (2) Pipette 1.2 mL of working solution into a cuvette (d = 1.0 cm) and equilibrate at 37°C for approximately 3 minutes. Add 0.04 mL of sample solution, mix and keep at 37°C for another 2 minutes.
- (3) Record the increase in optical density at 505 nm against water for 3 to 4 minutes in a spectrophotometer at 37°C and calculate the  $\Delta\text{OD}$  per minute from the initial linear portion of the curve ( $\Delta\text{OD}$  test). At the same time, measure the blank rate ( $\Delta\text{OD}$  blank) using the same method as

the test except that the enzyme diluent is added instead of the enzyme solution.

\*Dissolve the enzyme preparation in ice-cold enzyme diluent (**Reagent A**), dilute to **0.1–0.3 U/mL** with the same buffer, immediately before the assay.

- (4) Activity can be calculated by using the following formula:

**Volume activity (U/mL) =**

$$\frac{\Delta\text{OD}/\text{min} (\Delta\text{OD test} - \Delta\text{OD blank}) \times V_t \times df}{13.78 \times 1/2 \times 1.0 \times V_s}$$

$$= \Delta\text{OD}/\text{min} \times 4.499 \times df$$

**Weight activity (U/mg) = (U/mL) × 1/C**

Vt: Total volume (mL)

Vs: Sample volume (mL)

13.78: Millimolar extinction coefficient of quinoneimine dye under the assay conditions ( $\text{cm}^2/\text{micromole}$ )

1/2: Factor based on the fact that one mole of  $\text{H}_2\text{O}_2$  produces half a mole of quinoneimine dye.

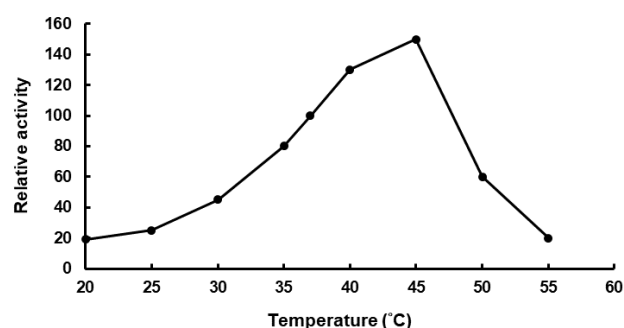
1.0: Light path length (cm)

df: Dilution factor

C: Enzyme concentration in dissolution (mg/ mL)

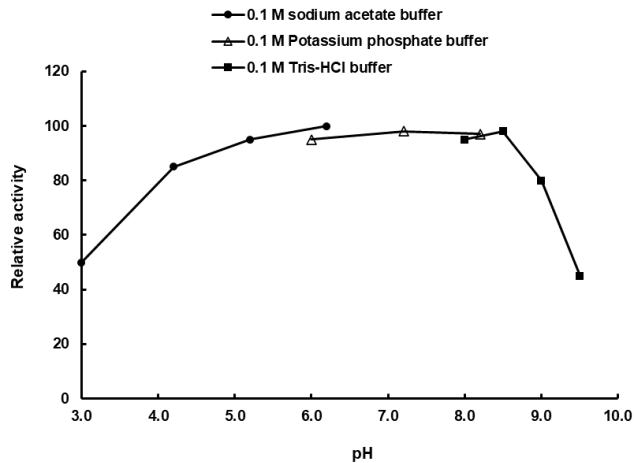
#### The effect of different conditions on Cholesterol esterase

A.



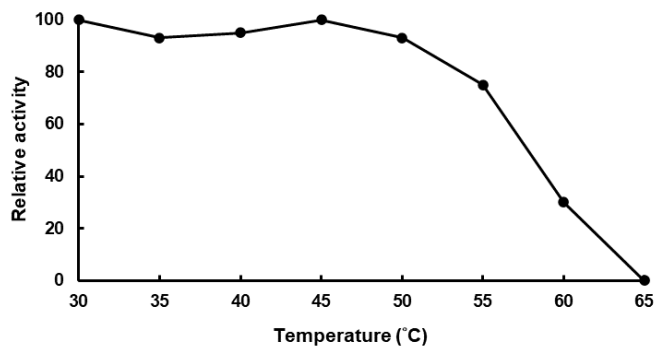
**Figure A. Temperature activity of Cholesterol esterase.** The enzyme reactions in 0.1 M Sodium acetate buffer, pH 5.5, were carried out under different temperatures.

B.



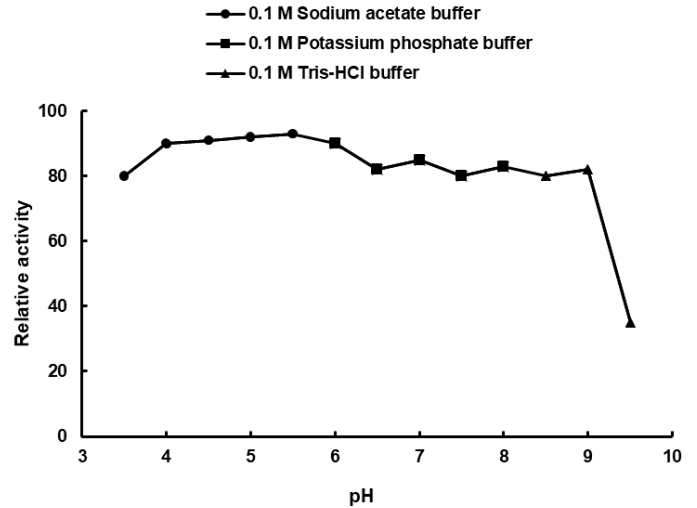
**Figure B. pH activity of Cholesterol esterase.** The buffer conditions with various pH values were used in the reaction.

C.



**Figure C. Thermal stability of Cholesterol esterase.** The enzyme powder was reconstituted by double-distilled water and treated with different temperatures for 15 minutes.

D.



**Figure D. pH stability of Cholesterol esterase.** The enzyme powder was reconstituted by double-distilled water and treated with different pH buffer conditions for 24 hours.

#### Disclaimer

For Research Use or Further Manufacturing Only

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